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5 *The Periodic Law*

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6 Chemical Bonding

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CHAPTER 6 REVIEW Chemical Bonding

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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered.

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Modern Chemistry 46 Chapter Test Chapter: Chemical Bonding In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

____ 1. The charge on an ion is a. always positive. b. always negative. c. either positive or negative. d. zero. ____ 2.

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CHAPTER 16 REVIEW Reaction Energy SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. For the following examples, state whether the change in entropy favors the forward or reverse reaction: forward reaction a. $\text{HCl(l)} \rightarrow \leftarrow \text{HCl(g)}$ reverse reaction b. $\text{C}_6\text{H}_{12}\text{O}_6\text{(aq)} \leftrightarrow \text{C}_6\text{H}_{12}\text{O}_6\text{(s)}$ forward reaction c. $2\text{NH}_3\text{(g)} \dots$

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SECTION 1 continued 6. Some binary compounds are ionic, others are covalent. The type of bond favored partially depends on the position of the elements in the periodic table.

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